

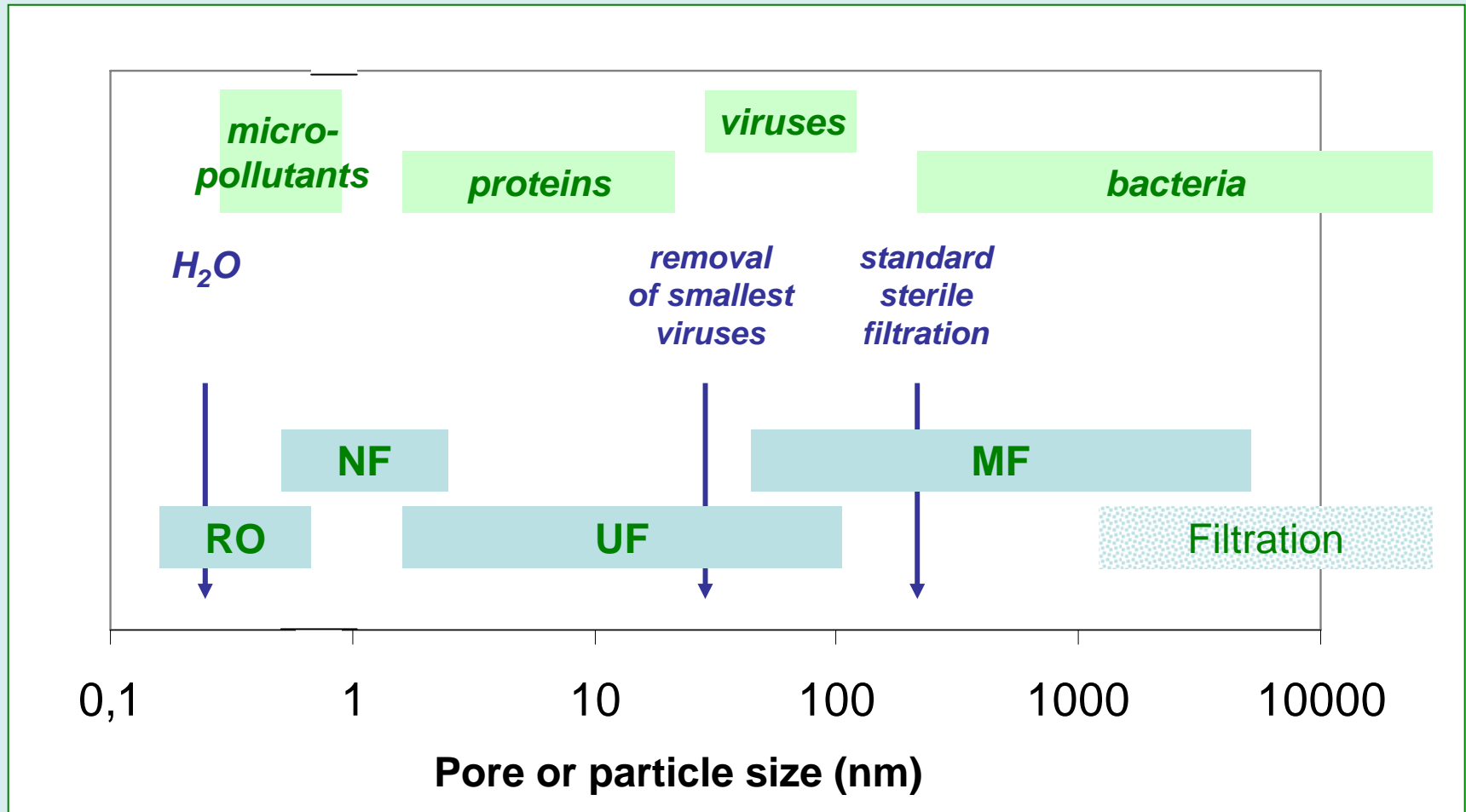
# Neue nanostrukturierte Membranen für die Wasseraufbereitung

**Mathias Ulbricht**

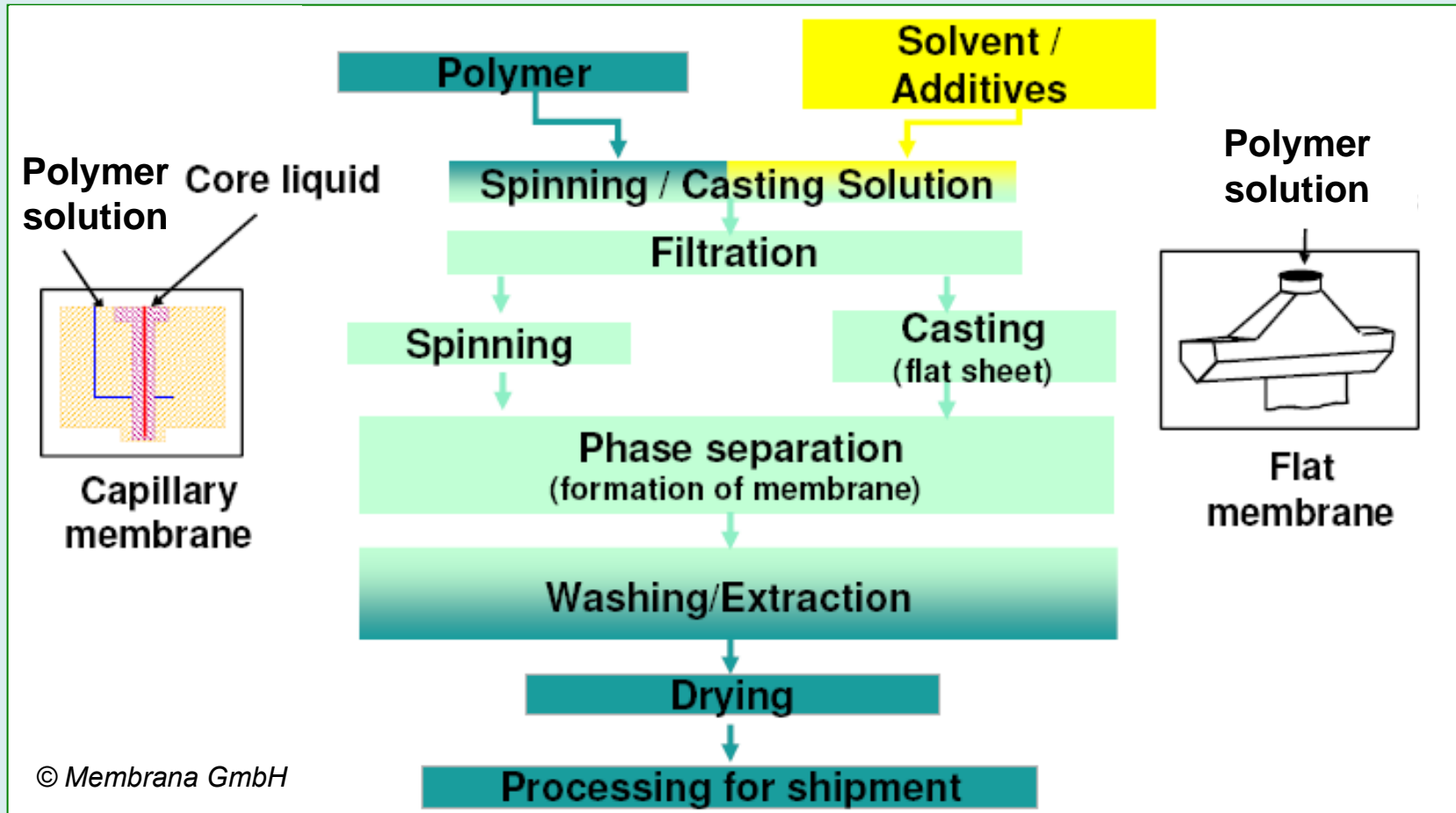
**Lehrstuhl für Technische Chemie II,  
Universität Duisburg-Essen, 45117 Essen, Germany**

**CeNIDE, 47057 Duisburg, Germany**

- ... pressure driven
- ... for water purification → „absolute barrier“ for various solutes / particles

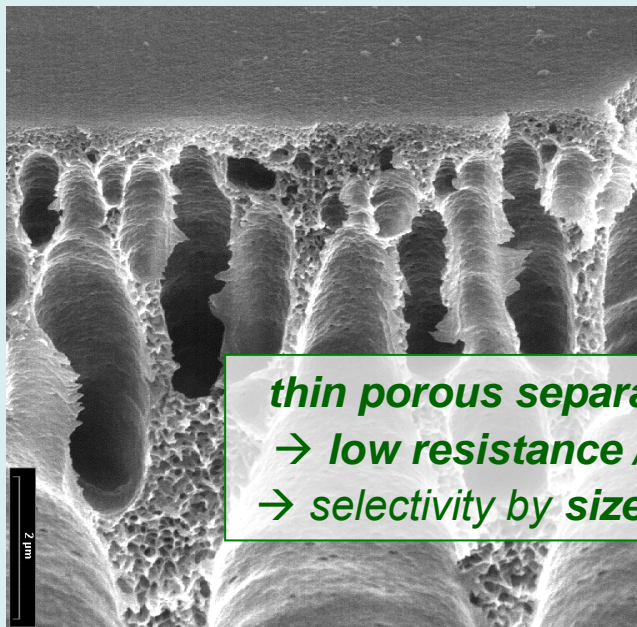


- ... via nonsolvent-induced phase separation (NIPS)
- ... established large-scale industrial processes



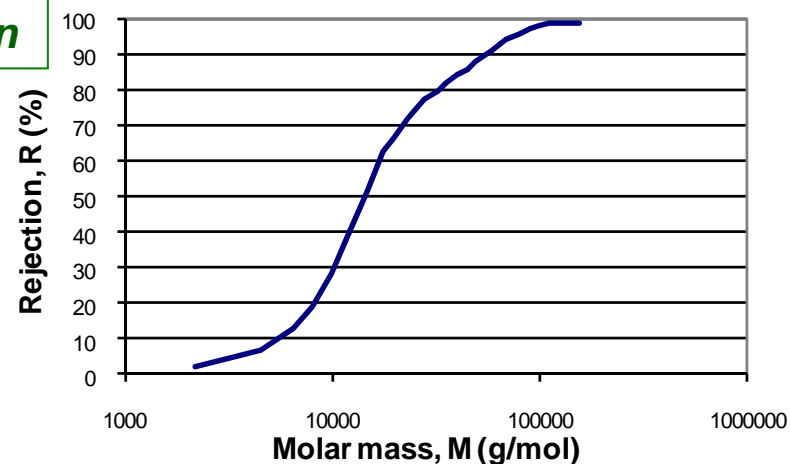
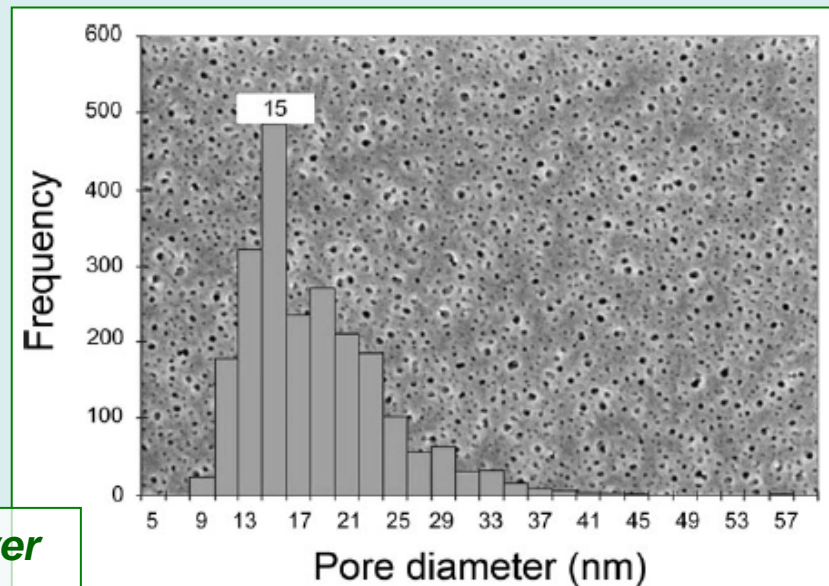
## Ultrafiltration membranes

here PAN; prepared by NIPS



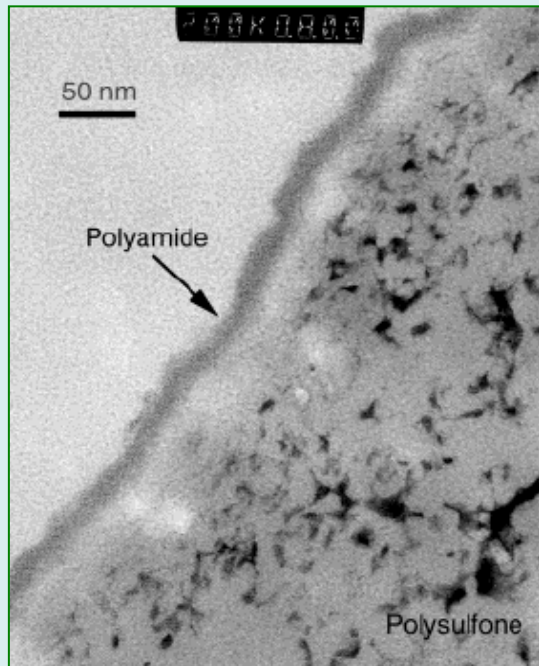
*thin porous separation layer*  
→ *low resistance / high flux*  
→ *selectivity by size exclusion*

- however:**
- *relatively broad pore size distribution*  
→ *moderate size selectivity*
  - *relatively low porosity*  
→ *limited flux*

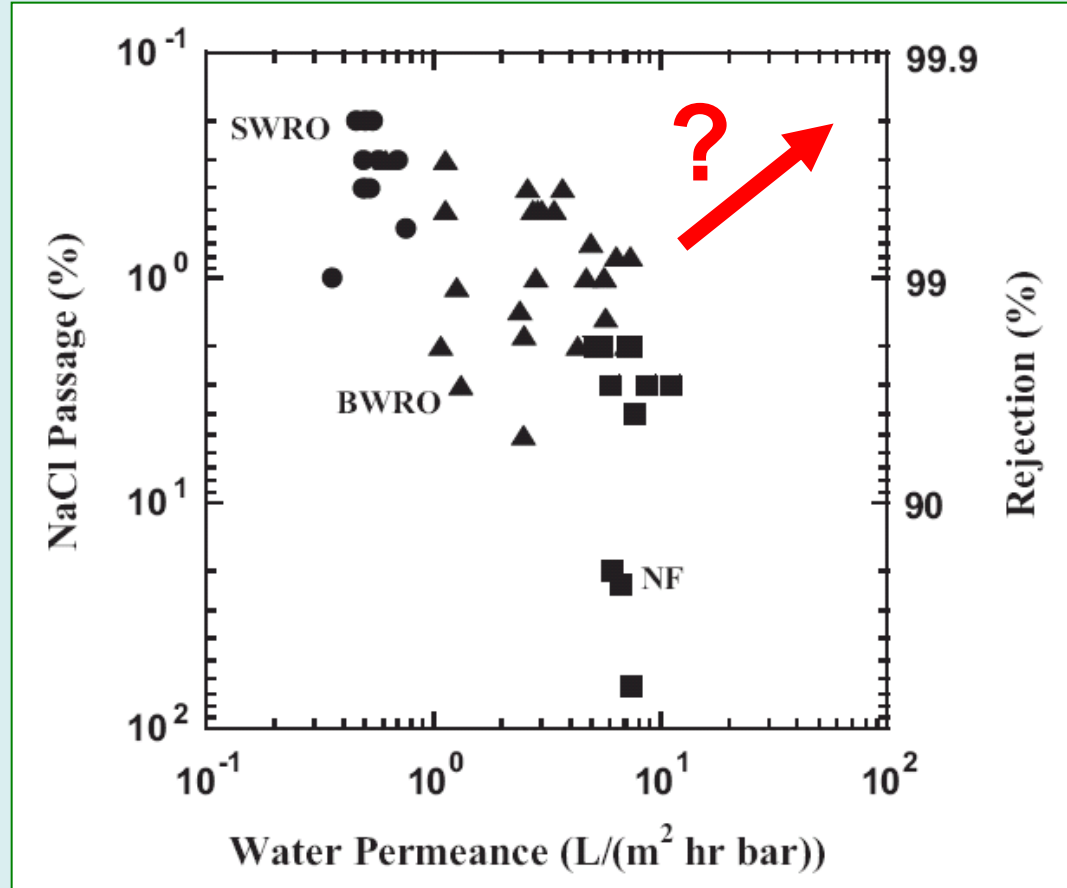


## Reverse osmosis membranes

polyamide, prepared by interfacial polymerization on UF-type support membrane



Thin barrier layer of RO composite membrane.



Trade-off between permeability and selectivity for state-of-the-art polyamide composite RO membranes.

## Reverse osmosis for desalination:

well-defined objective for separation = **pure H<sub>2</sub>O**

easy standardization (compared to other water purifications!)

## Aims

- higher permeability
- higher selectivity (especially for uncharged solutes!)
- higher stability
- minimize concentration polarization (*i.e.*, *accumulation of rejected solute*)
- no fouling (*i.e.*, *reversible or irreversible deposition on the membrane*)
- minimize cleaning

→ smaller foot-print

→ less material

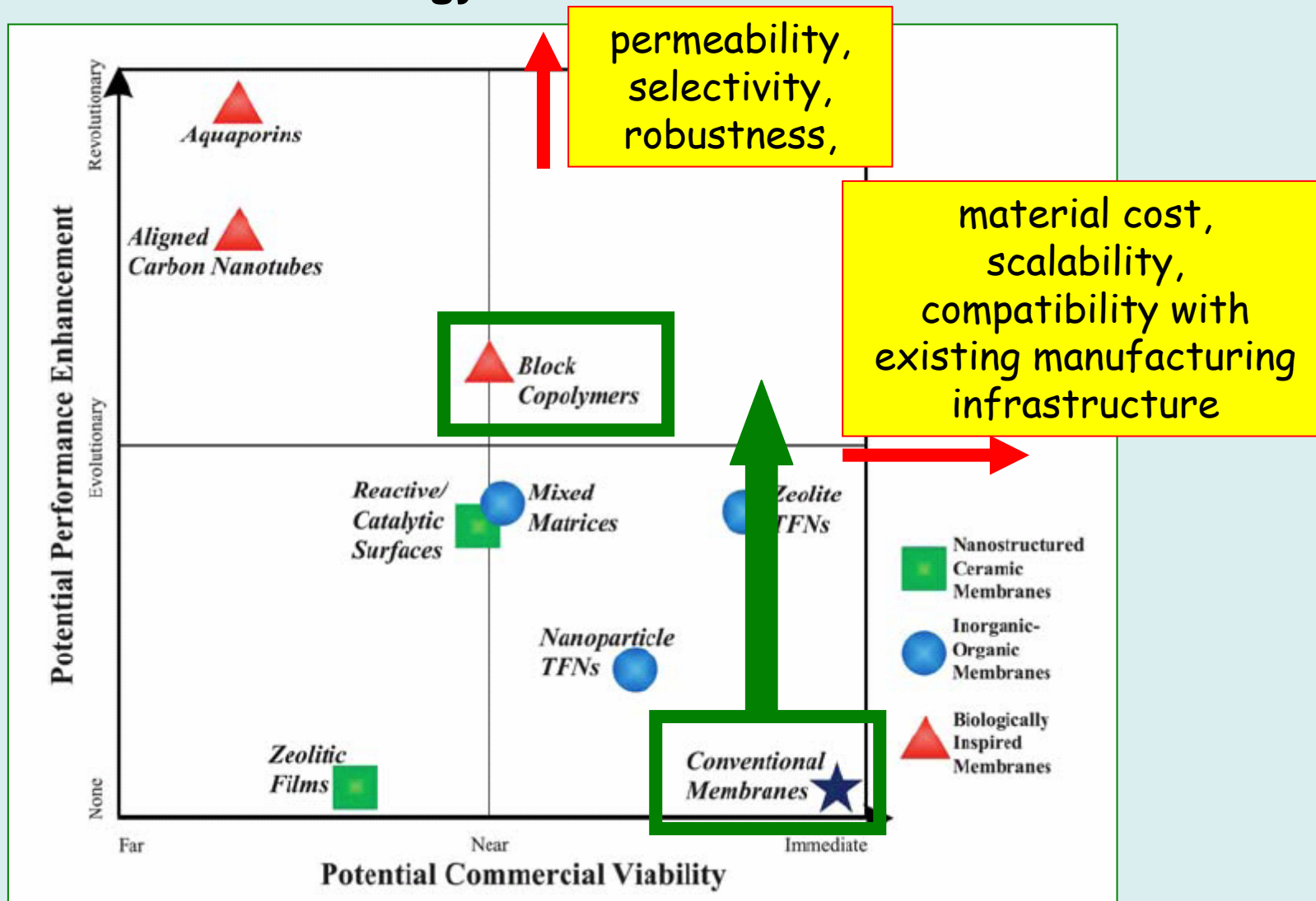
→ less energy

→ less/no cleaning chemicals

→ lower investment cost

→ lower operating cost

... via nanotechnology



## Aims for improvement

(for all membrane processes!)

- higher permeability
- higher selectivity (especially for uncharged solutes!)
- higher stability
- minimize concentration polarization
- no fouling
- minimize cleaning

} → *these problems are also, typically even more relevant for membranes with high permeability!*

## One (also our) strategy

add or integrate additional functions to/in membrane barrier structure

**by functional nanolayers or nanoparticles**

→ protection\*

→ adsorber

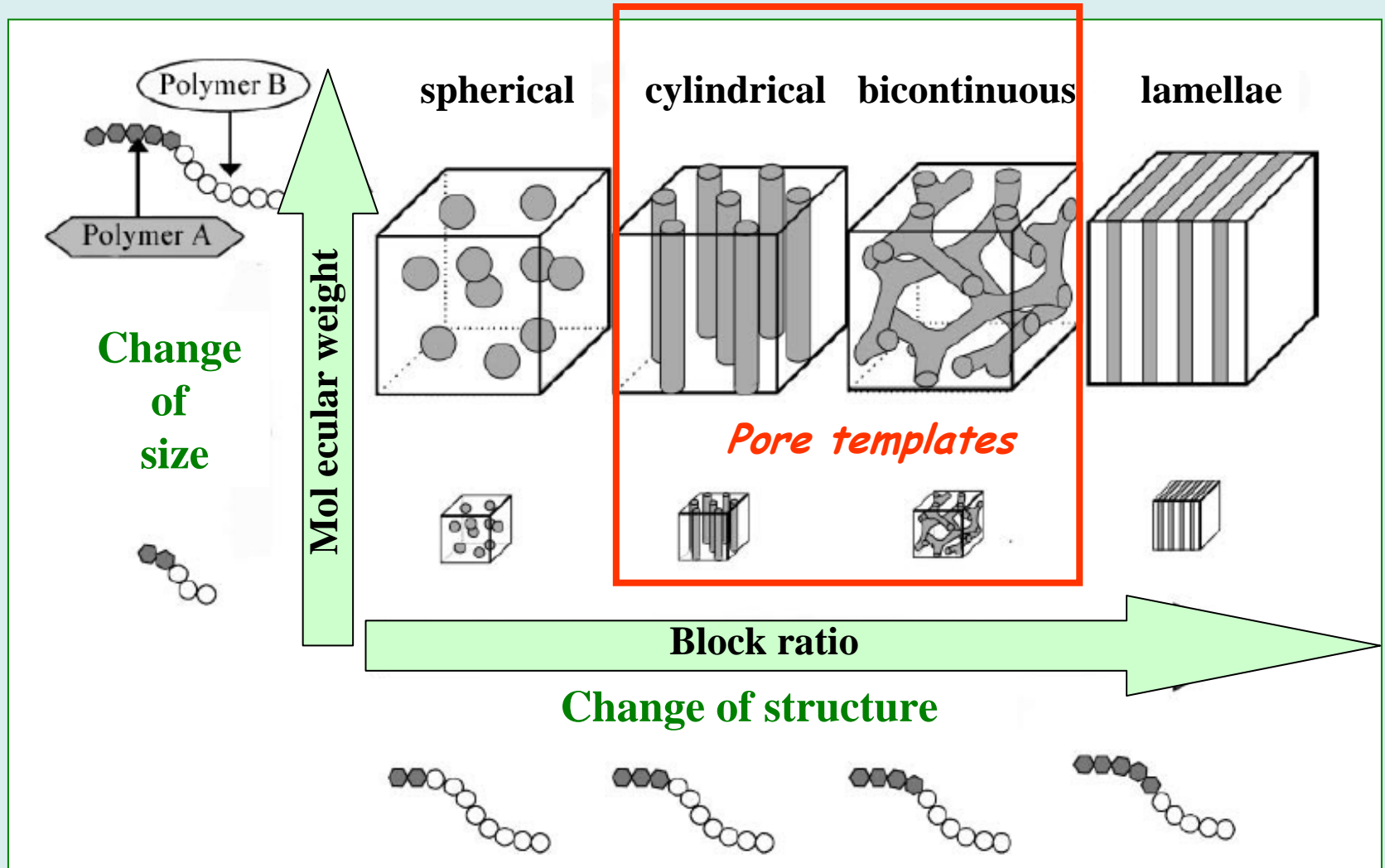
→ catalyst\*

→ stimuli-responsivity\*

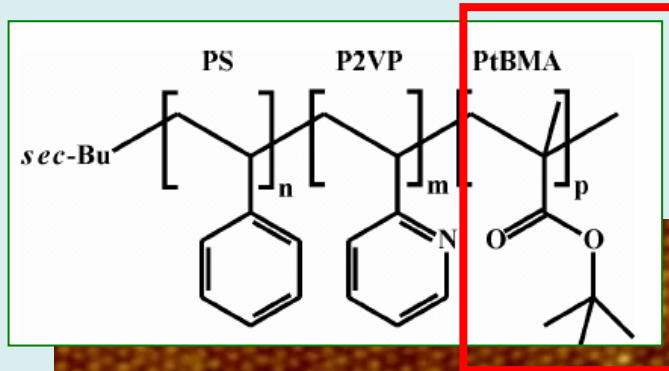
} → additional elimination pathways,  
e.g., for micropollutants

\* → anti-fouling / fouling-release

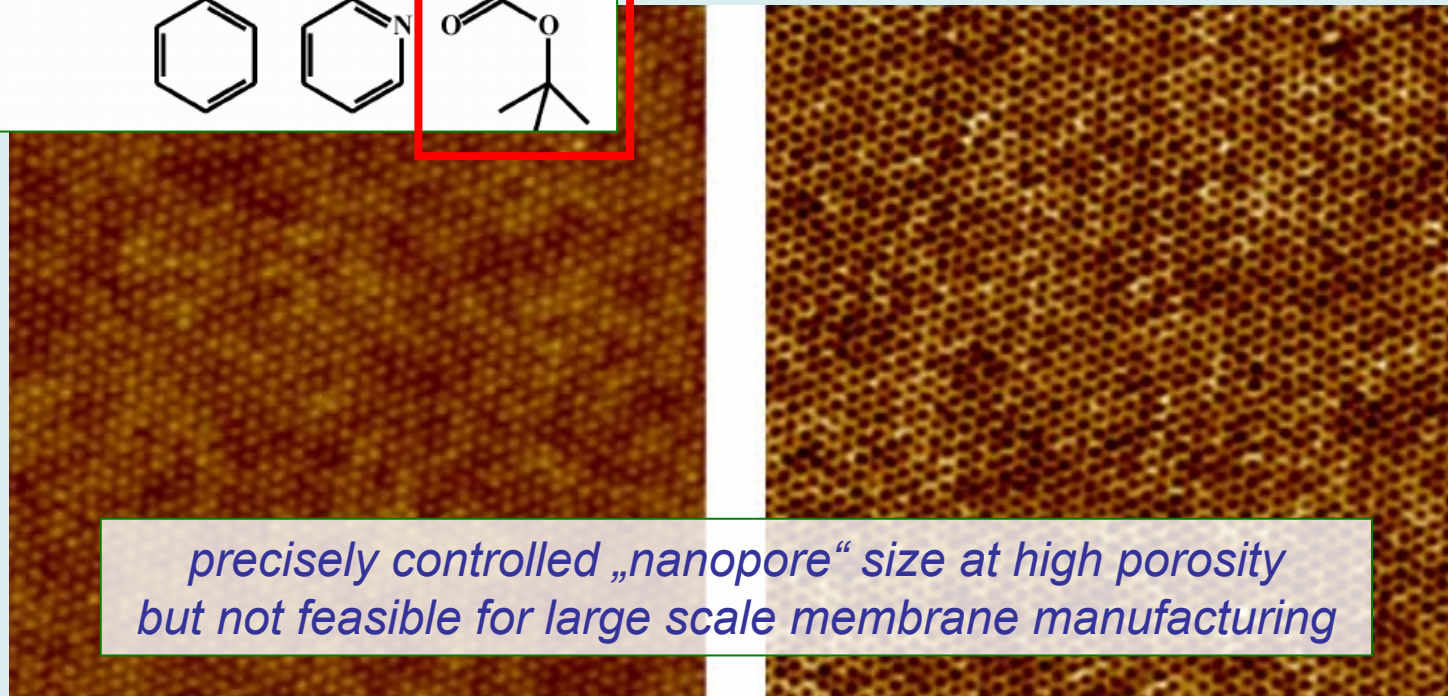
... tailoring pore and surface structure on the nanoscale



## Phase separated block copolymers as pore template



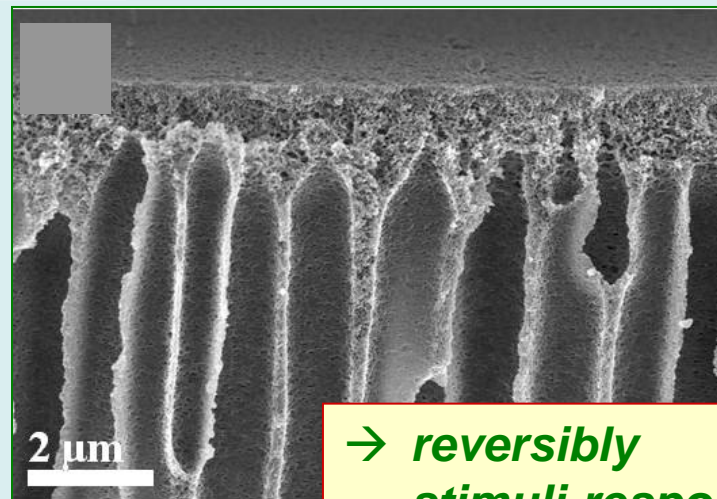
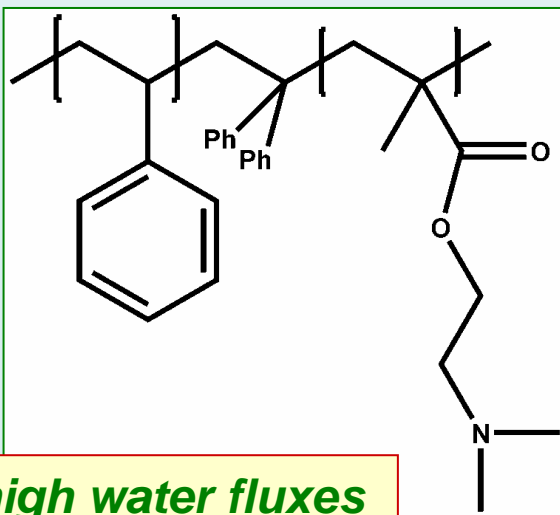
*block (microdomain)  
can be removed by UV irradiation*



*precisely controlled „nanopore“ size at high porosity  
but not feasible for large scale membrane manufacturing*

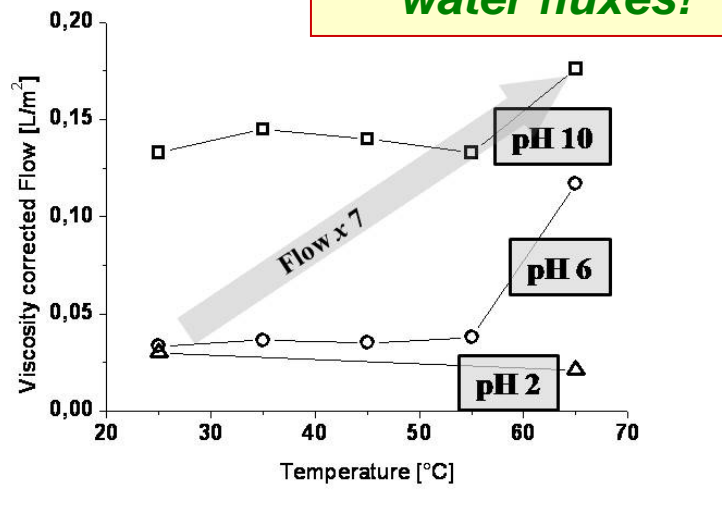
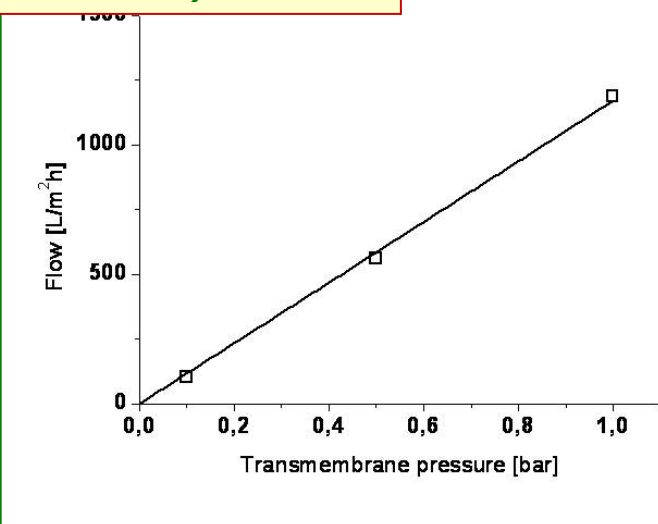
tapping mode AFM topography of triblock copolymer film:  
before ... after **UV irradiation**

... via scalable NIPS process

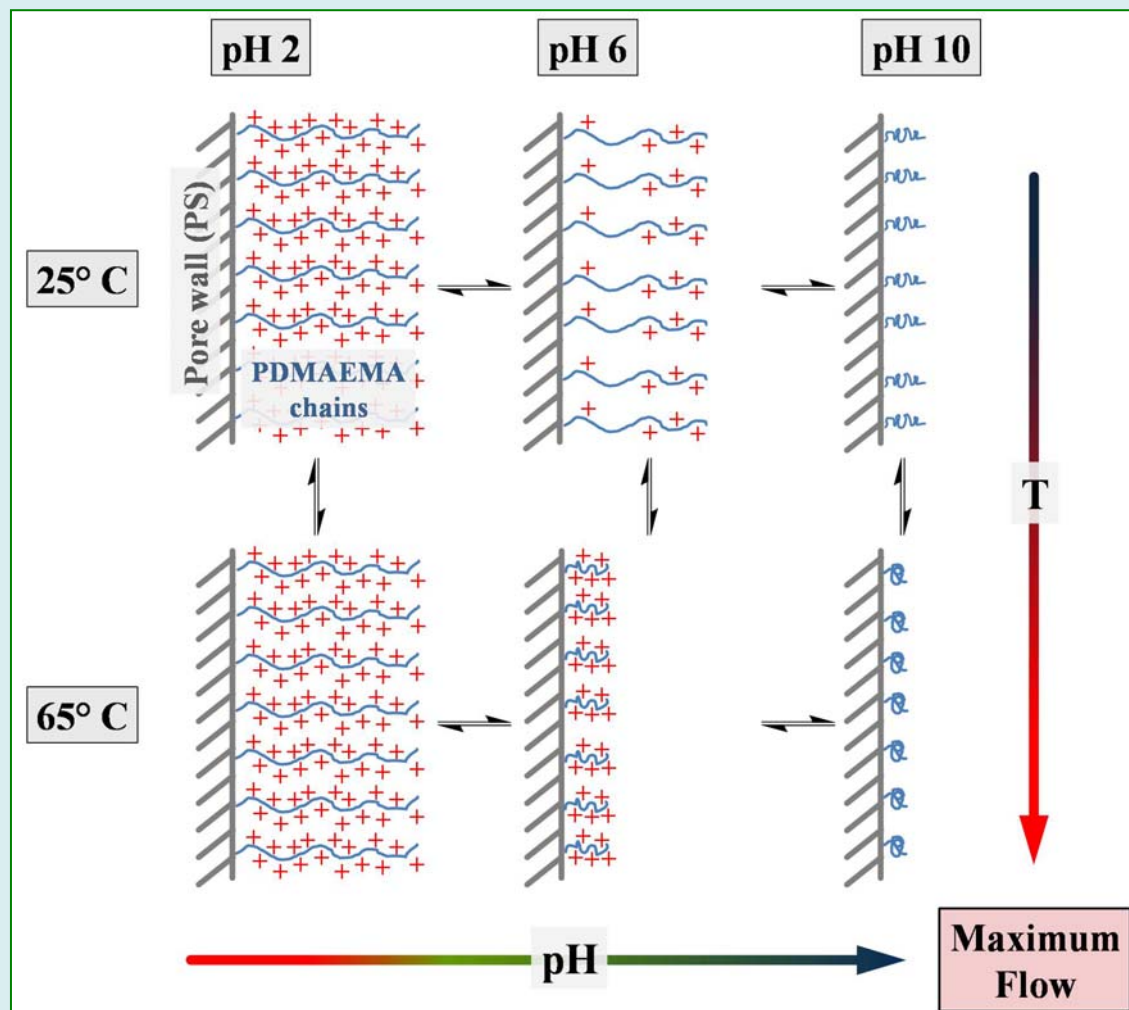


→ high water fluxes  
(UF ... MF)

→ reversibly  
stimuli-responsive  
water fluxes!



## Phase separated, double stimuli-responsive pore structure

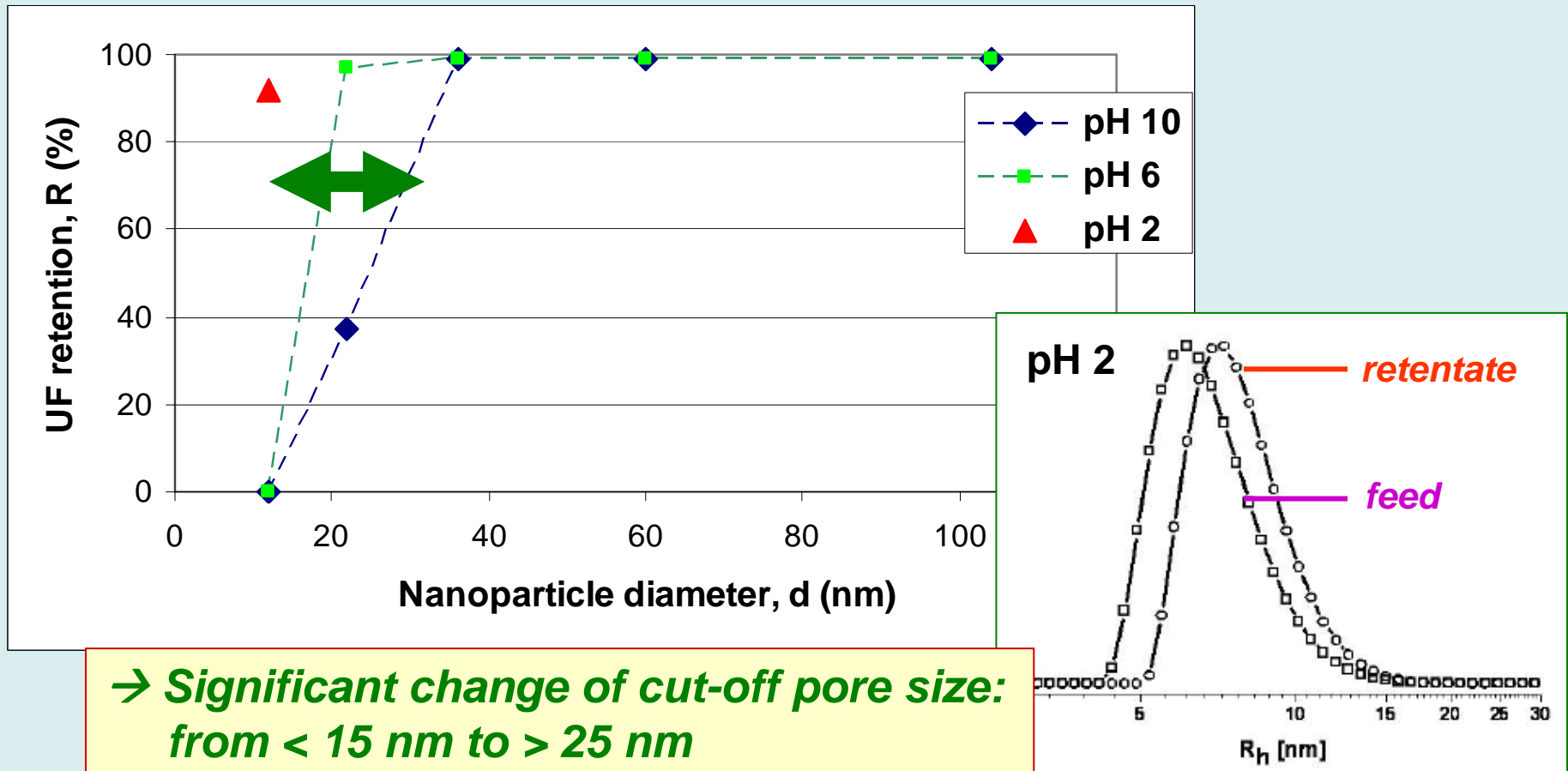


*Surface and pore structure from micro- and macropore separation*

## Stimuli-responsive ultrafiltration of nanoparticles

as function of pH (@ 25°C)

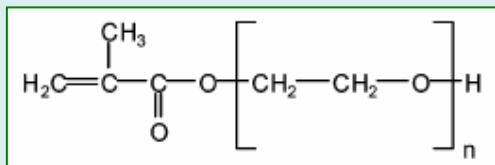
silica (22 to 104 nm), LUDOX AM-30 (12 nm), 0.1 wt% in water, 0.2 bar.



## Graft copolymerization of anti-fouling layers on UF membranes

UV wavelength: > 300 nm

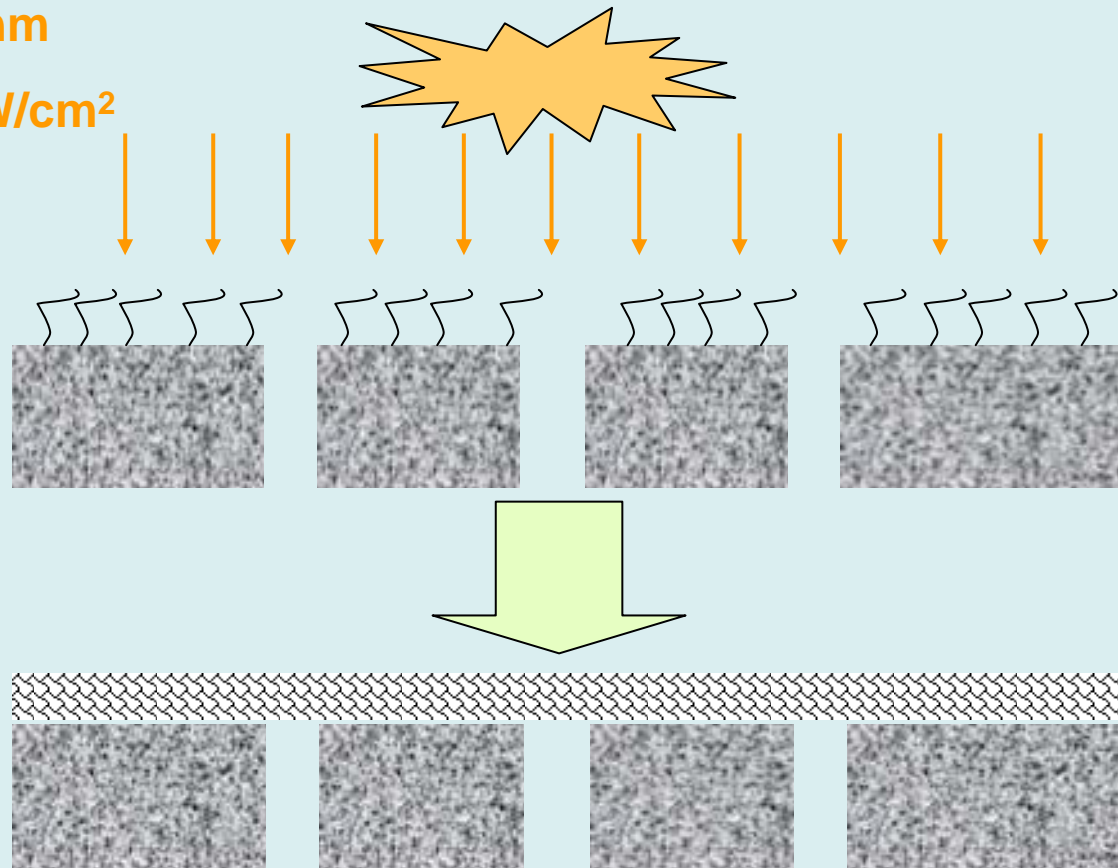
UV intensity:  $35 \pm 5$  mW/cm<sup>2</sup>



monomer (PEGMA)

PES UFM (100 kDa)

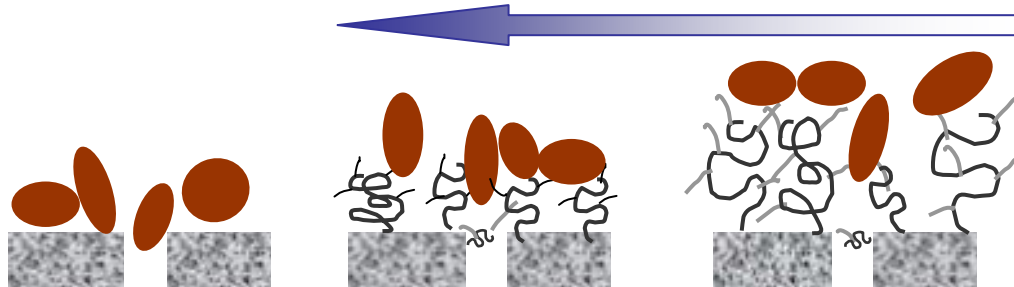
grafted ultrathin  
hydrophilic, flexible  
polymer layer



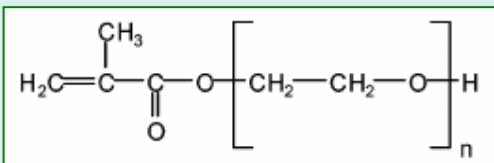
**PES-based thin-layer hydrogel composite membrane**

... via grafted polymer hydrogel layers

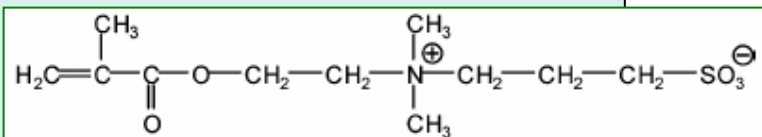
→ high fouling tendency  
hydrophobic („*clustered water*“)



Hydrophilic,  
water-swollen (~„*free water*“)  
→ fouling-resistant



PEGMA



SPE

Grafted polyPEGMA vs. polySPE (zwitterionic)      Equilibrium swelling ratio

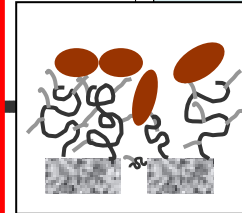
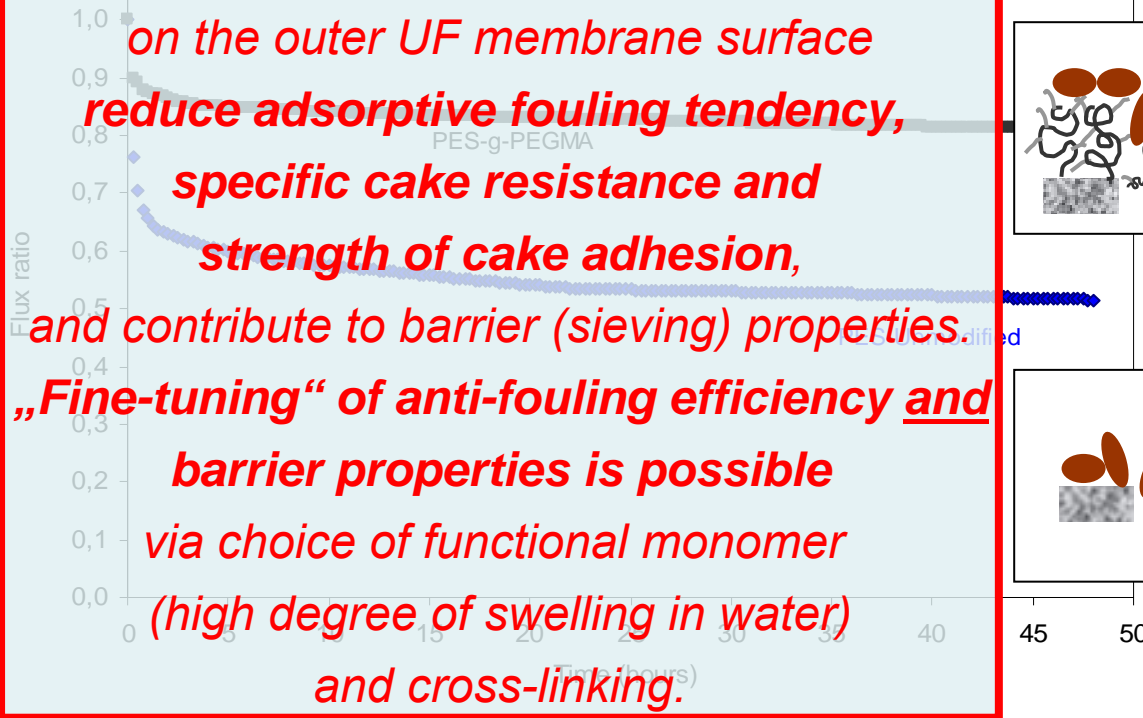
at same DG:      flux      PEGMA > SPE  
                         rejection      PEGMA < SPE

polySPE	7.7
polyPEGMA	12.2

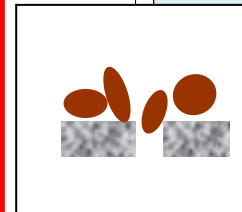
fouling resistance PEGMA > SPE

## PES-based thin-layer hydrogel composite membrane (MWCO 10 kDa)

*Thin-layer hydrogels as „protective“ layer on the outer UF membrane surface reduce adsorptive fouling tendency, specific cake resistance and strength of cake adhesion, and contribute to barrier (sieving) properties. „Fine-tuning“ of anti-fouling efficiency and barrier properties is possible via choice of functional monomer (high degree of swelling in water) and cross-linking.*



PES-g-PEGMA



PES unmodified

**- similar results with proteins, polysaccharides and mixtures thereof.**

H. Susanto, M. Ulbricht, *Langmuir* **2007**, 23, 7818-7830.

H. Susanto, M. Ulbricht, *Water Research* **2008**, 42, 2827-2835.

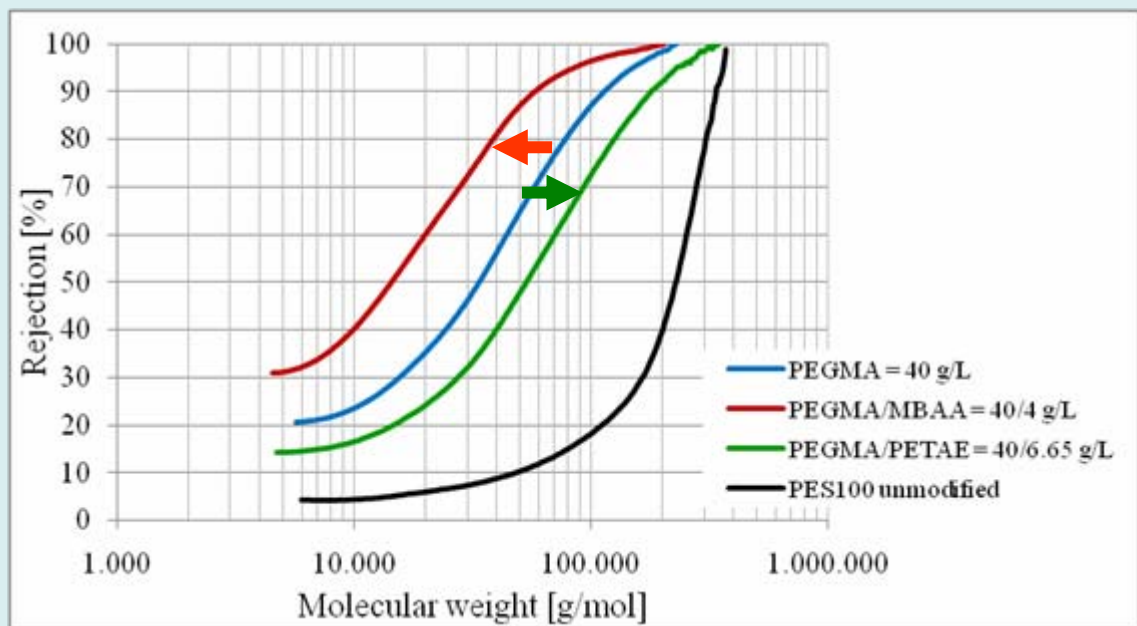
H. Susanto, H. Arafat, E. Janssen, MU, *Separ. Purif. Tech.* **2008**, 63, 558.

# PES-based thin-layer hydrogel composite membrane

PES UF membrane (100 kDa, Sartorius) photo-grafted with polyPEGMA

→ influence of cross-linker structure and ratio

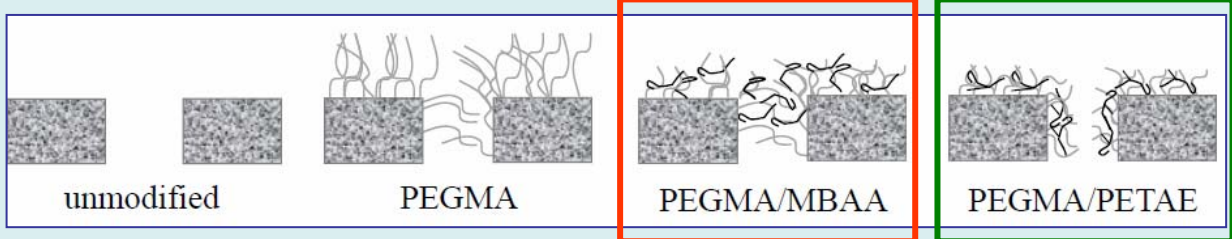
sieving curves (dextran)



additional sieving through hydrogel

pore covering hydrogel

PES 100 kDa  
high cross-linker content  
UV dose 11 mJ/cm<sup>2</sup>



## ... from surface grafting to thin-layer composite membranes

common sense: kosmotropic surfaces (PEG, zwitterionic, ...) are best suited, but there are also specific requirements:

with respect to foulants:

**biofouling**

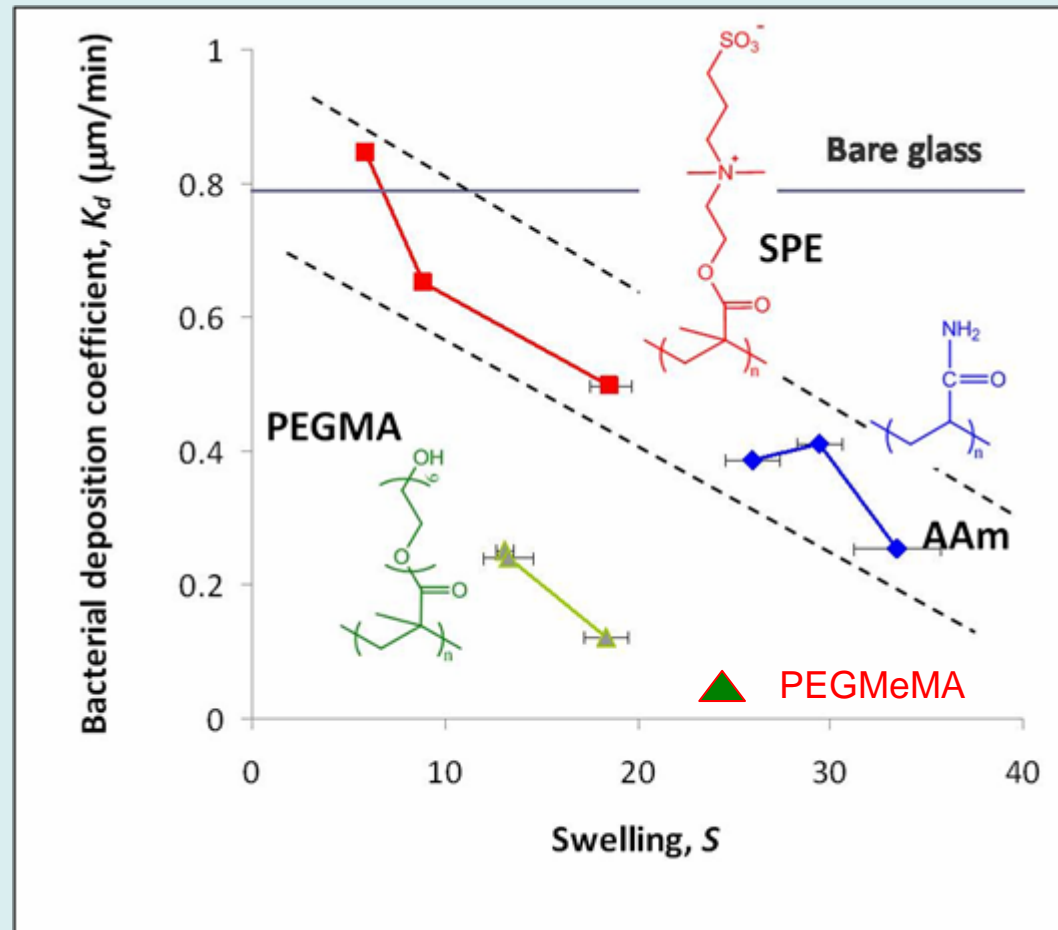
→ maximize  
swelling degree

**organic fouling**

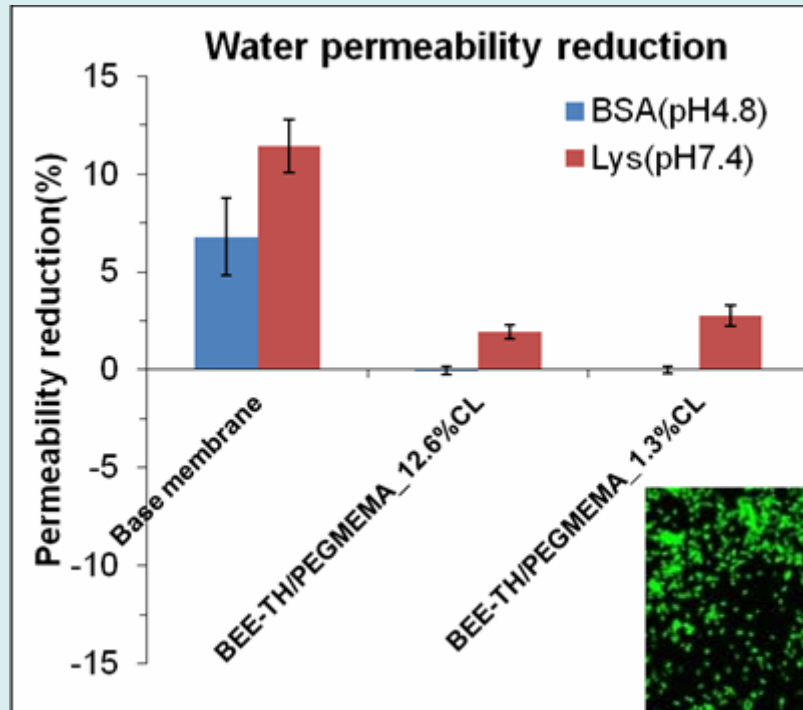
→ exclude  
biomacromolecules

+ **mechanical strength**

Initial deposition of  
*P. fluorescens* F113  
on different hydrogels  
at varied cross-linking



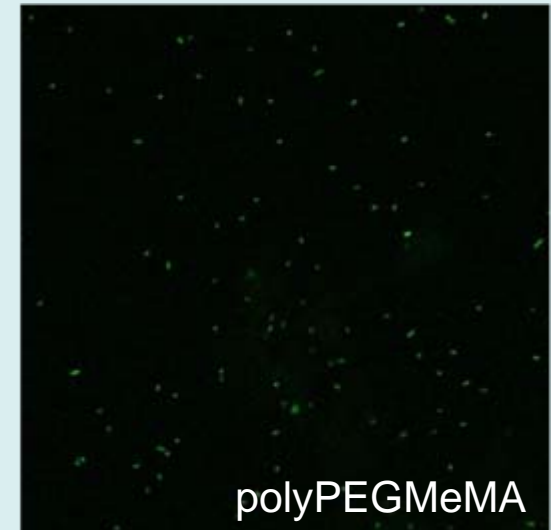
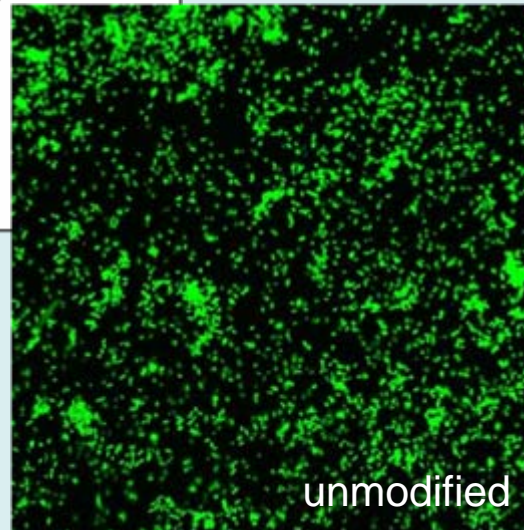
... on polyamide composite NF membranes (NF 270)  
after protein filtration (1 hour)



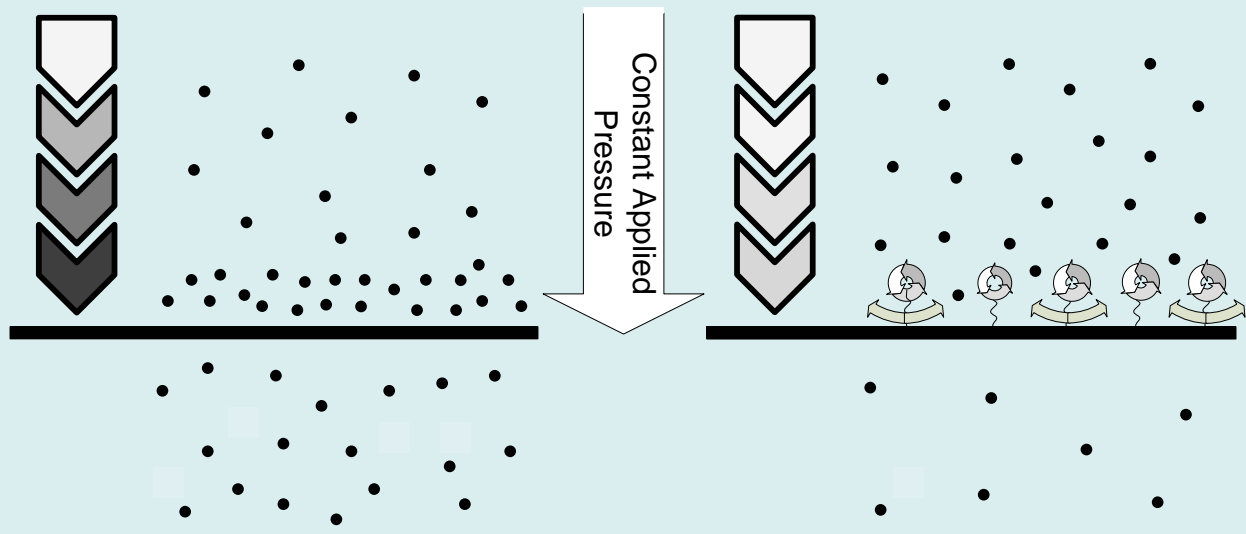
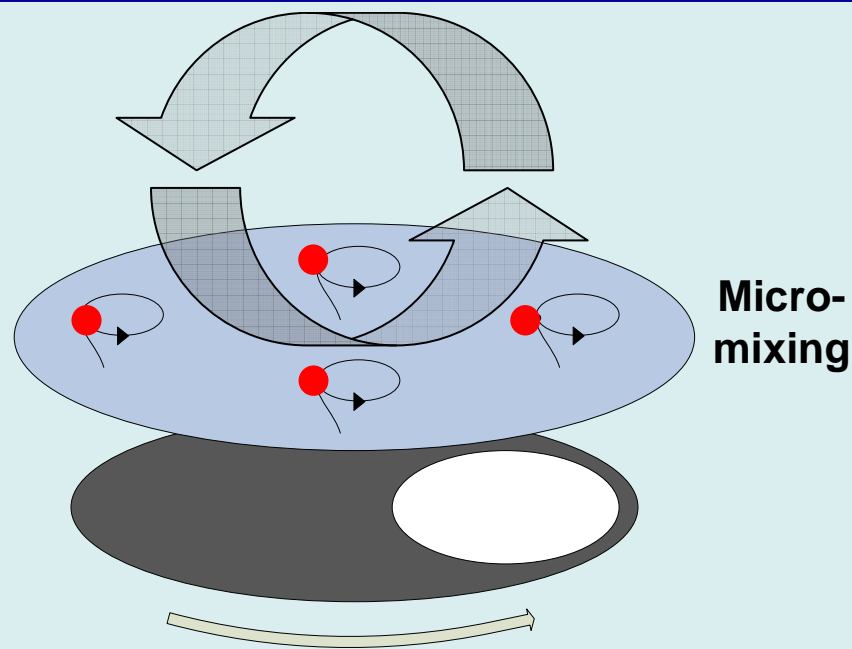
→ *largely reduced organic fouling (but function of solute size!)*

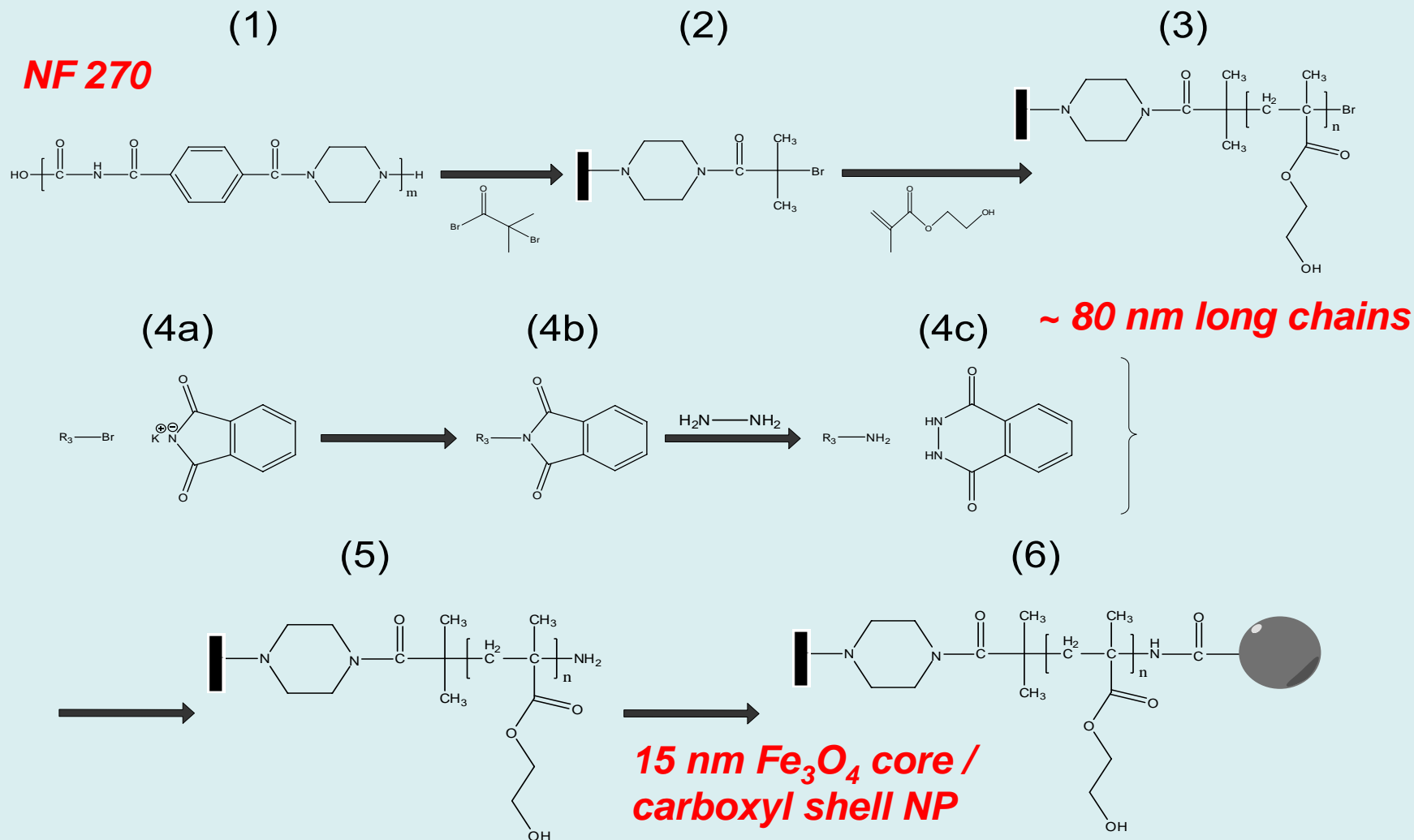
after biofilm growth  
(*P. fluorescens* F113; 2 days)

→ *almost eliminated biofouling*



## Magnetically activated micromixers for separation membranes

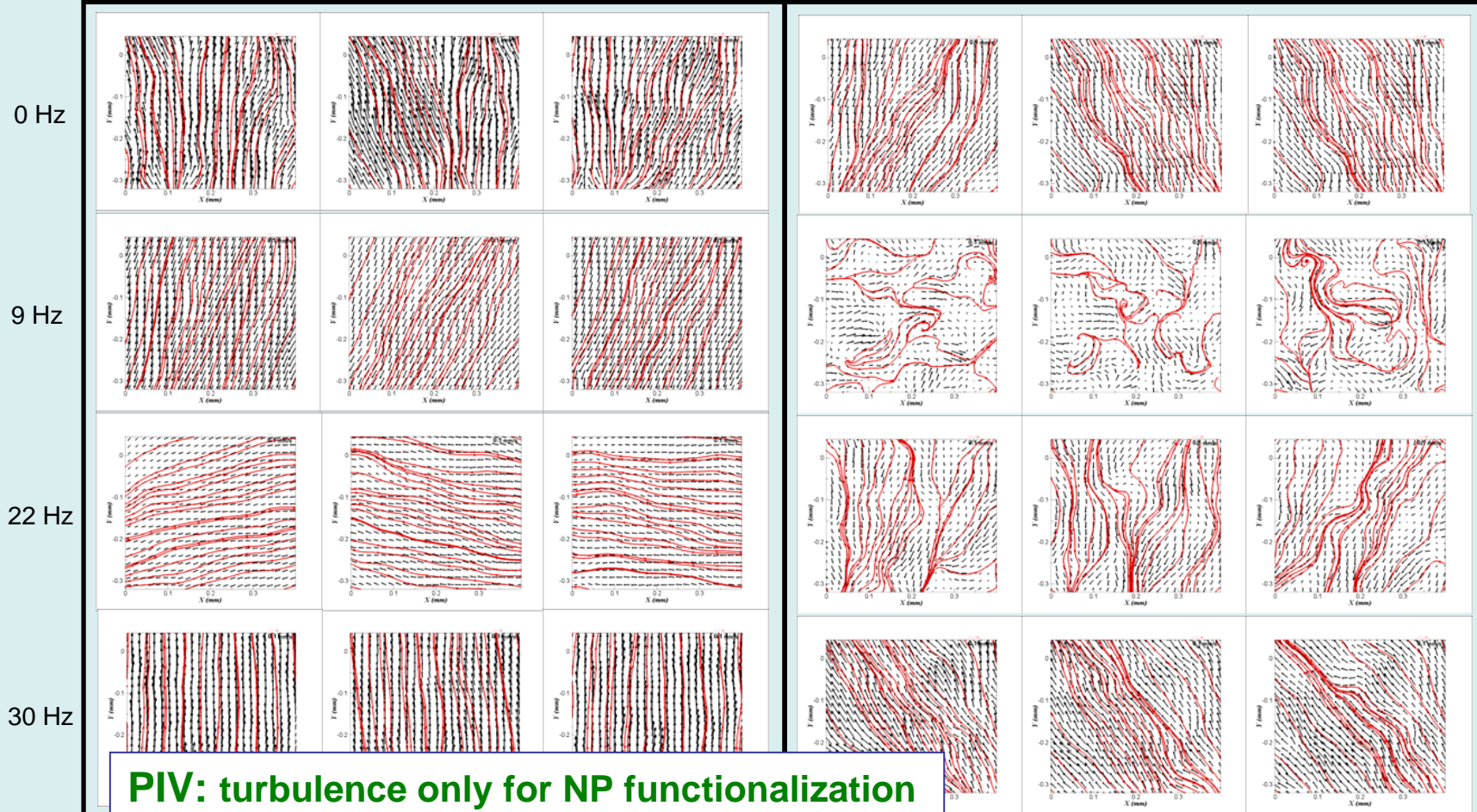


Synthesis by controlled surface initiated graft copolymerization  
and nanoparticle coupling

## Visualization of macromixing

Base membrane (NF 270)

NP functionalized NF 270



**PIV: turbulence only for NP functionalization and medium rotation frequency!**

## Stimuli-responsive nanofiltration membrane

### micromixing at the surface reduces concentration polarization

Table 1. Average Fluxes and Salt Rejections for Control and Modified Membranes at 45 psig (3.1 bar)

	500 ppm CaCl <sub>2</sub>		2000 ppm MgSO <sub>4</sub>	
	control	modified	control	modified
Salt Rejection (%)				
with field	34.4 ± 0.2	<u>40.4 ± 0.2</u>	66.5 ± 0.2	<u>74.4 ± 0.2</u>
without field	32.5 ± 0.2	34.2 ± 0.2	66.0 ± 0.2	67.7 ± 0.2
Filtrate Flux (L/m <sup>2</sup> ·h)				
with field	13.4 ± 0.6	<u>9.6 ± 0.7</u>	5.4 ± 0.3	<u>6.0 ± 0.4</u>
without field	12.8 ± 0.6	7.8 ± 0.7	5.6 ± 0.4	4.0 ± 0.2

**NF: significant improvement of flux and salt rejection only for NP functionalization and rotating magnetic field (~ 10 Hz)**

**Polymeric membranes for RO, NF or UF are well established materials; function and performance depend critically on their nanoscale structure (the same is true for inorganic membranes).**

**Current performance limitations can be overcome by advanced nanotechnologies.**

**Surface functionalization with „protective“ (anti-fouling, fouling-release) nanolayers is most relevant (and already done at industrial scale); careful adaptation to the respective application and membrane barrier structure are necessary.**

**Integration of adsorber or catalytic properties via functional nanoparticles into separation membranes is a parallel approach.**

**Next generation membranes will comprise:**

- stimuli-responsive surfaces or pores**
- functional nanochannels (e.g., carbon nanotubes)**

*... to the group*

*... to collaborators & their group:*

*Prof. Dr. R. Wickramasinghe  
Colorado State University, U.S.A.*

*Prof. Dr. V. Freger  
Ben-Gurion University, Israel*

S07

Fakultät für Chemie



Essen, June 2011.

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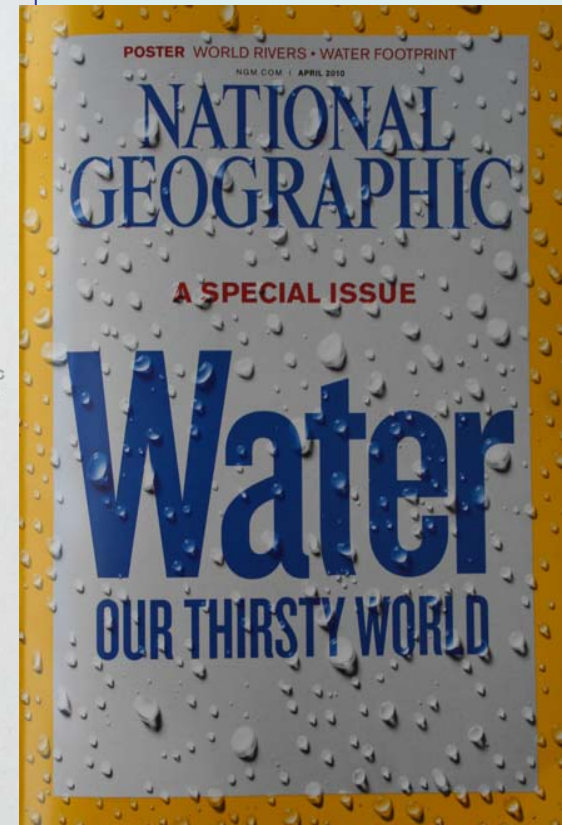
# Bio-inspired nanotechnology: maximize selectivity at high permeability

**Three technologies** promise to reduce the energy requirements of desalination by up to 30 percent. The race is on to see which will take the lead.

**FORWARD OSMOSIS**  
Water molecules migrate by natural osmosis, without energy input, into an even more concentrated "draw solution," whose special salt (green) is then evaporated away by low-grade heat.  
**On the market: 2010-2012**

**CARBON NANOTUBES**  
An electric charge at the nanotube mouth repels positively charged salt ions. The uncharged water molecules slip through with little friction, reducing pumping pressures.  
**On the market: 2013-2015**

**BIOMIMETICS**  
Water molecules pass through channels made of aquaporins, proteins that efficiently conduct water in and out of living cells. A positive charge near each channel's center repels salt.  
**On the market: 2013-2015**



April 2010